

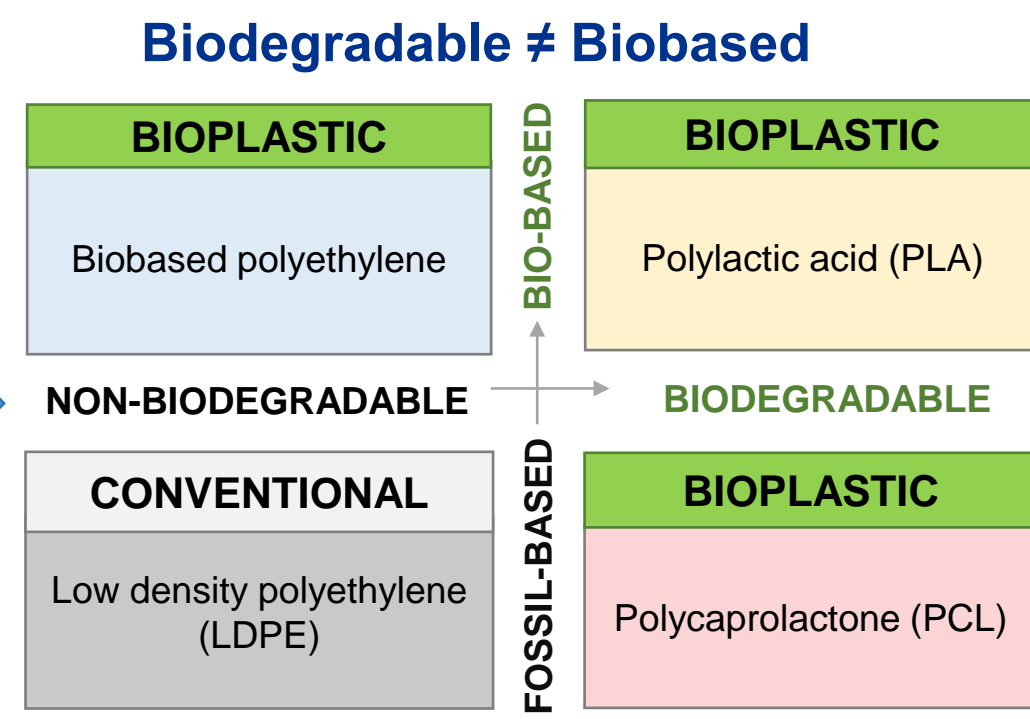
# PLASTIC POLLUTION: CAN THE CHEMISTRY OF MICROPLASTICS INFLUENCE THEIR ENVIRONMENTAL IMPACT?

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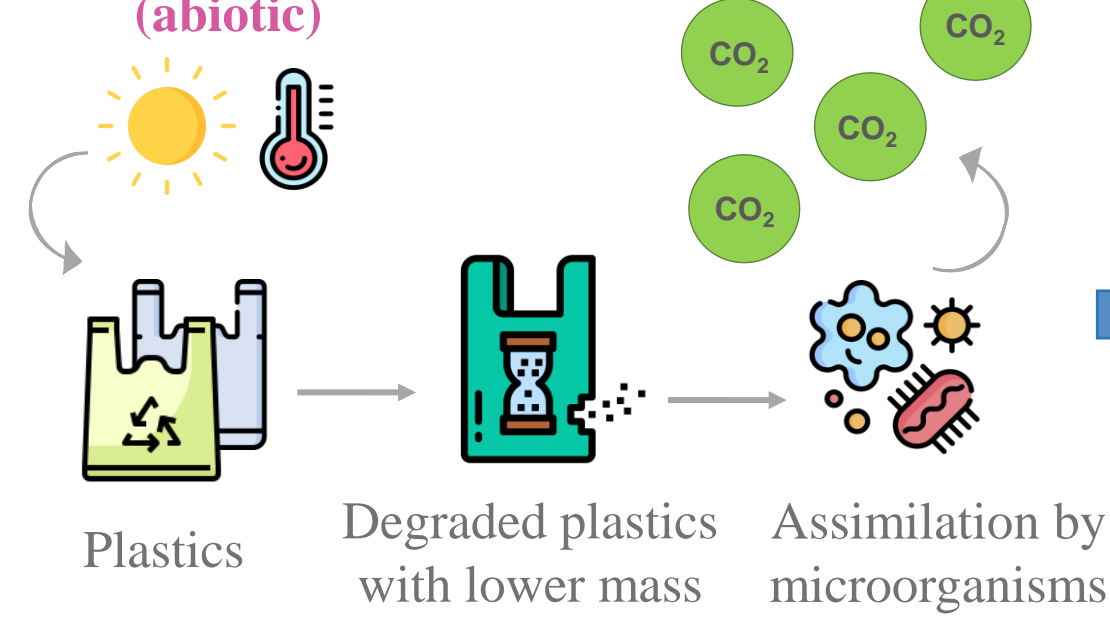
## 1. The Challenge



Plastic is everywhere



### Environmental factors (abiotic)



Can we mimic the formation of microplastics in the laboratory?  
**Artificial weathering!**

## 2. Research objectives and methodology

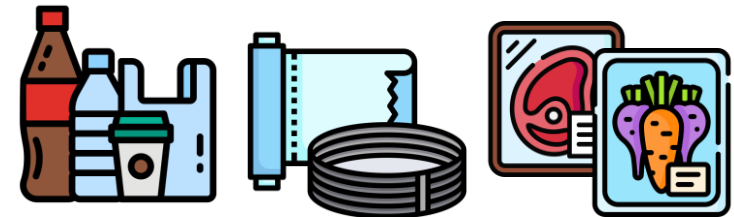
**Ob1:** Artificial weathering of plastics and characterisation

**Ob2:** Link between weathering and chemical structure

**Ob3:** Environmental impact of artificial plastic fragments

### a. Plastics with different chemistry

Polyethylene is the most ubiquitous type of plastics in the environment due to its use:



**LDPE**  
Fossil-based and non biodegradable

**OXO LDPE**  
Fossil-based with degradation catalyst

### b. Accelerated artificial weathering

**SUNLIGHT - PHOTO DEGRADATION**

- Q-Lab QUV<sup>®</sup> Accelerated Weathering tester
- Main degradation factor in the environment
- Degradation continues if sunlight is removed

**HEAT - THERMAL DEGRADATION**

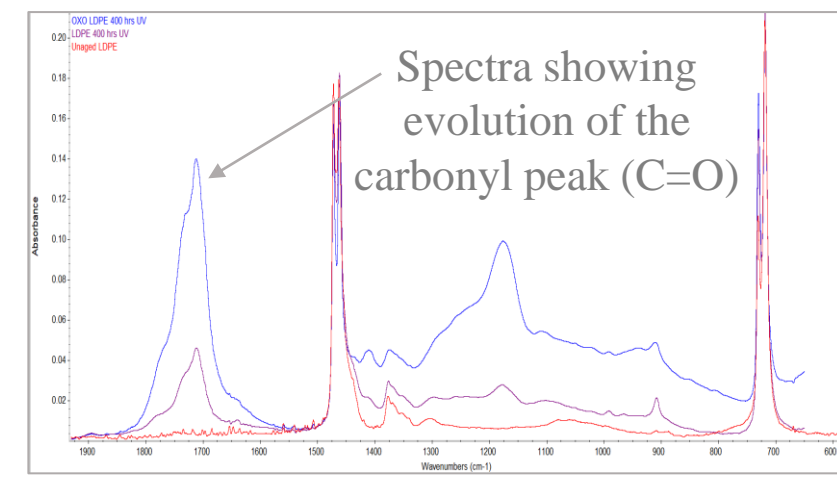
- Oven (50-70 °C)
- Influences secondary photochemical reactions
- Degradation rate is temperature-dependent

### c. Characterisation of polymer degradation

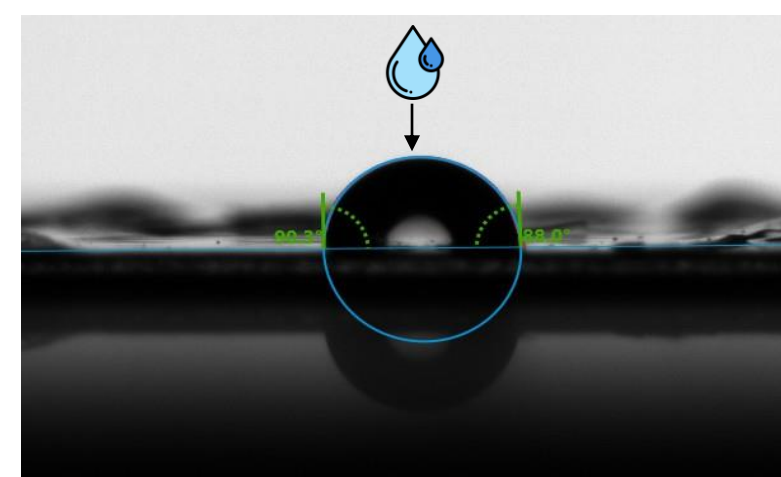
Preparation of unaged and aged plastic samples



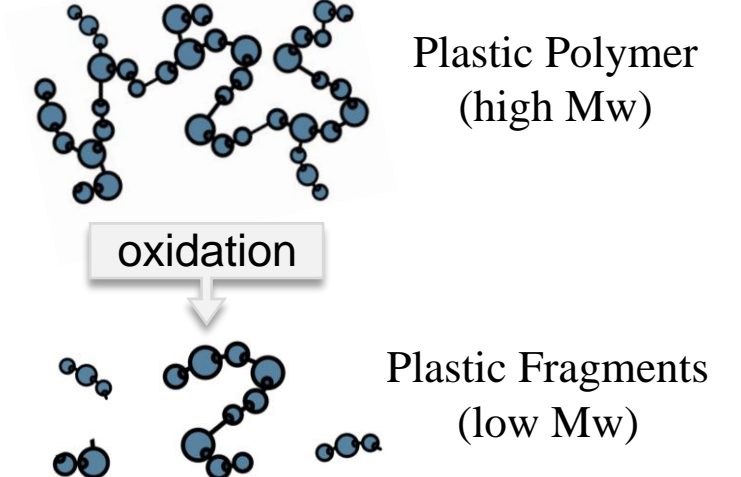
Infrared spectroscopy to quantify oxidation



Changes to surface hydrophobicity by contact angle

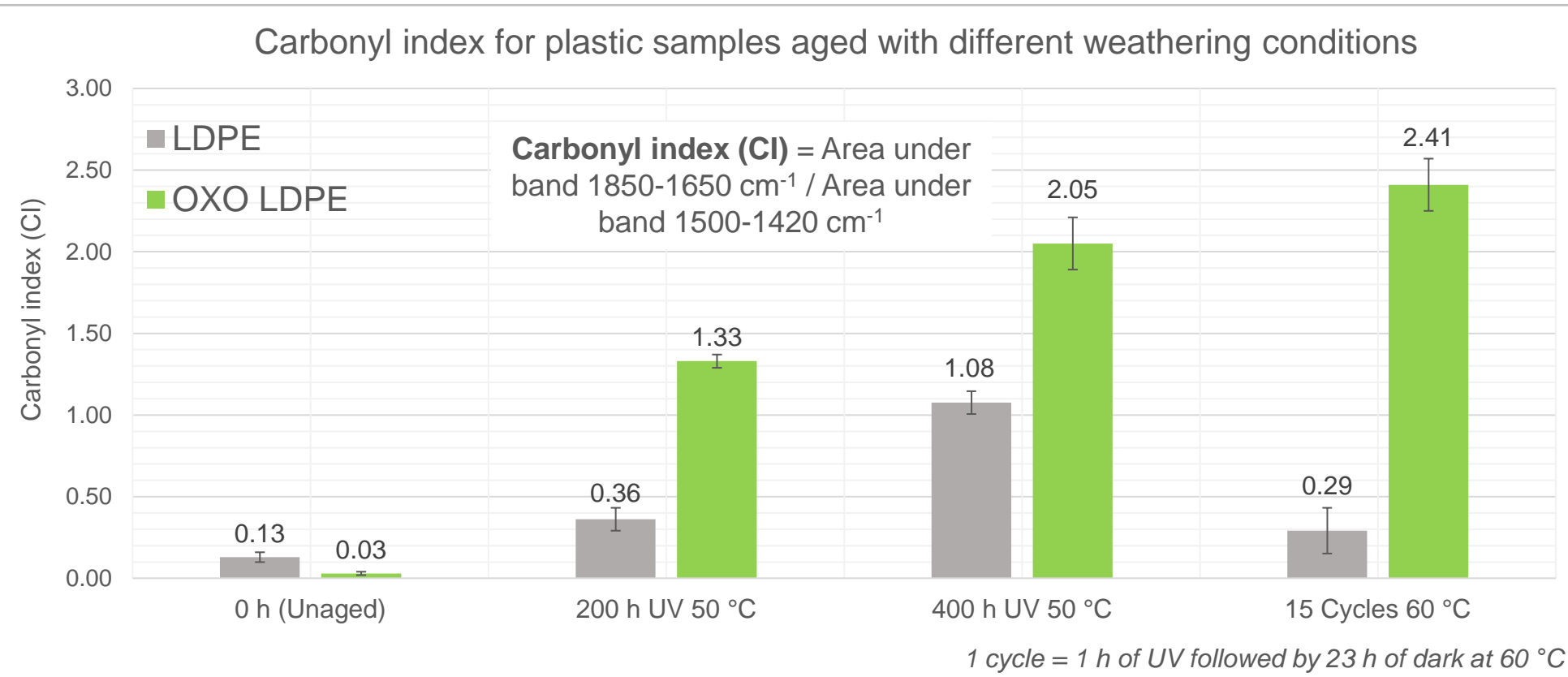


Decrease of molecular weight by gel permeation chromatography



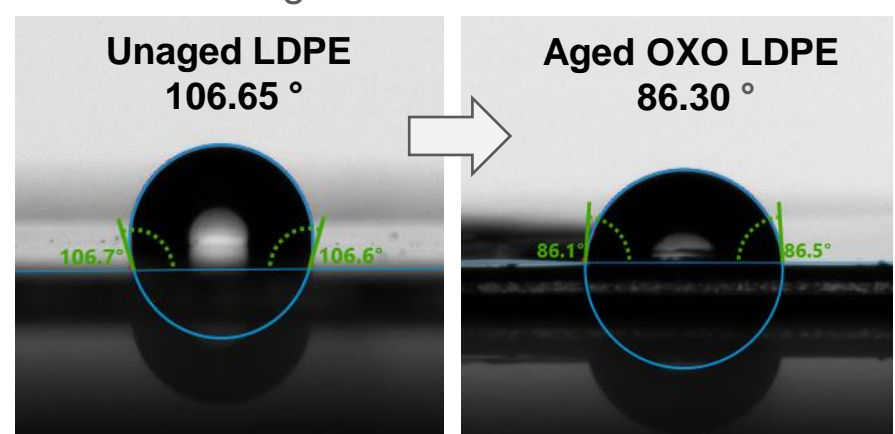
## 3. Artificial weathering: LDPE vs. OXO LDPE

**Method:** Plastic samples subjected to different ageing conditions, via a combination of photo-oxidation (UV) and thermo-oxidation (heat) at different temperatures and different times.



Carbonyl index data provide evidence that different ageing protocols have varying effects depending on the type of plastic and weathering.

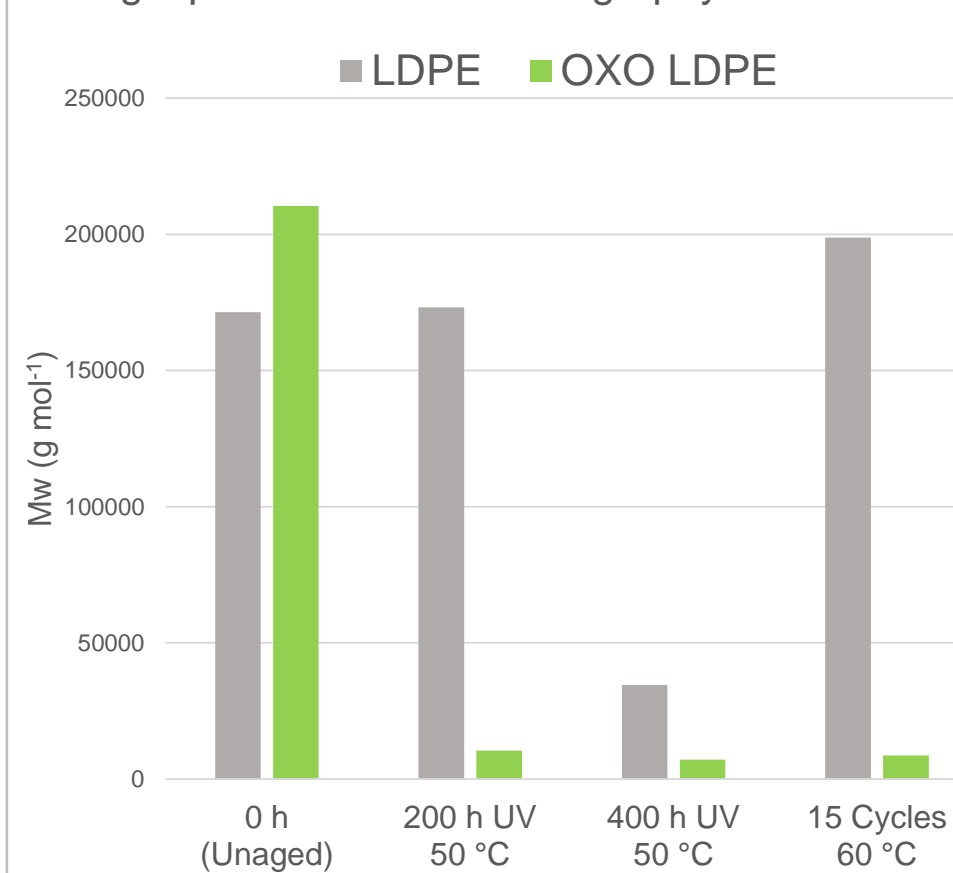
Changes in surface hydrophobicity by contact angle:



Formation of hydrophilic carbonyl groups on the surface decrease the contact angle

CA (°)	0 h Unaged	200 h UV 50 °C	400 h UV 50 °C	15 Cycles 60 °C
LDPE	106.65	104.50	97.90	106.52
OXO LDPE	104.64	95.19	74.95	78.75

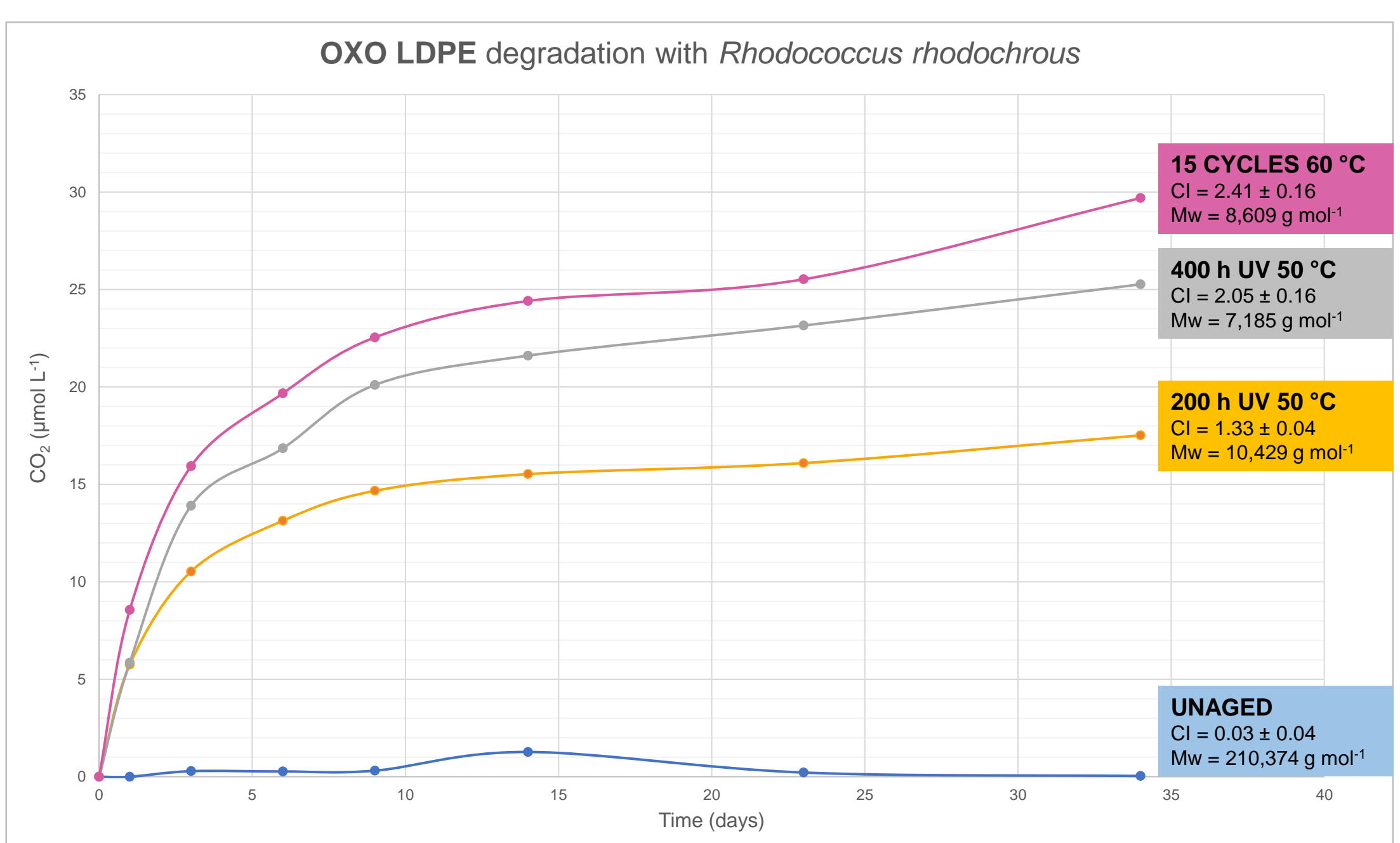
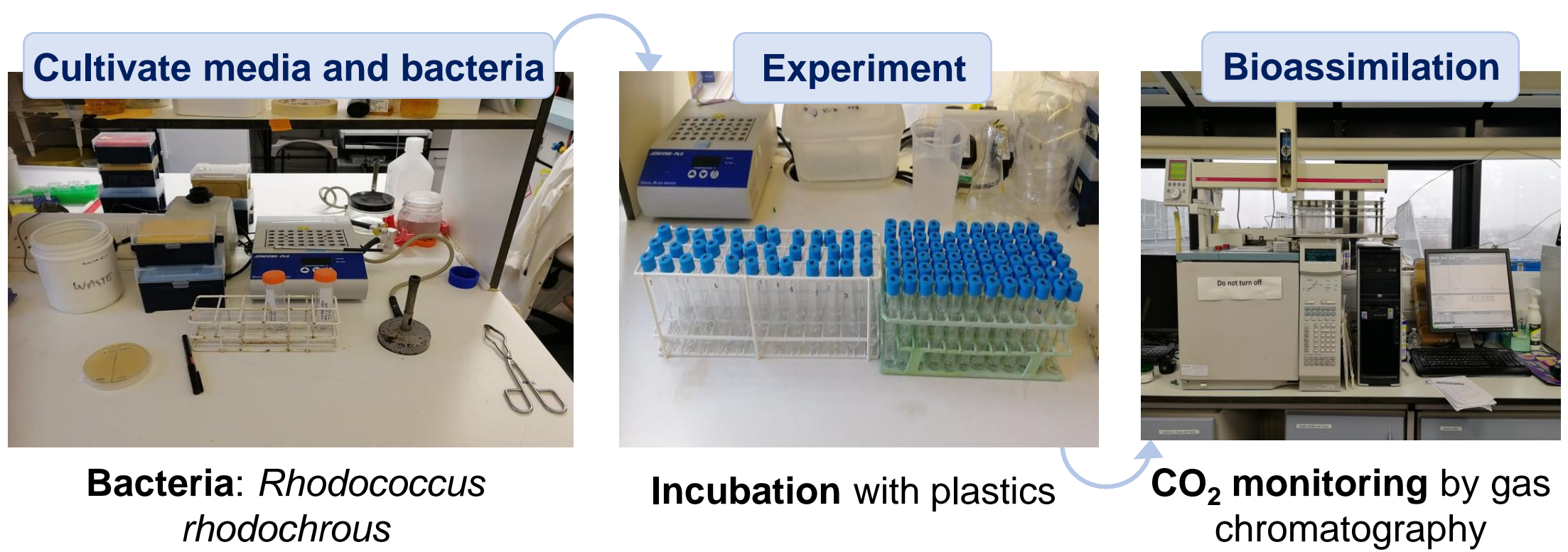
Changes in molecular weight determined by gel permeation chromatography:



The presence of a degradation catalyst in OXO LDPE results in fragments with different chemical structure and morphology being formed with artificial weathering.

## 4. Environmental impact of plastic fragments

**Method:** Plastic samples incubated with *Rhodococcus rhodochrous*. Graph demonstrates the amount of CO<sub>2</sub> released by the bacteria depending on plastic type and ageing conditions as a result of microbial assimilation.



OXO LDPE (15 Cycles 60 °C) shows higher amount of CO<sub>2</sub> release. Unaged OXO LDPE shows the least.

## 5. Conclusions

The combination of photodegradation (UV) and thermal degradation (heat) can be used successfully for artificial weathering.

Different plastics respond differently to ageing protocols, leading to fragments with specific chemistry.

*Rhodococcus* differentiates plastic types according to weathering conditions.

The environmental impact of plastic fragments will be the result of close interdependence between abiotic and biotic degradation mechanisms.

### References

Jambeck, et al. (2015). Moore, et al. (2008). Thompson, et al. (2009). BSI PAS 9017:2020. Almond, et al. (2020). European Bioplastics (2019). Song, et al. (2009). Lucas, et al. (2008).

