



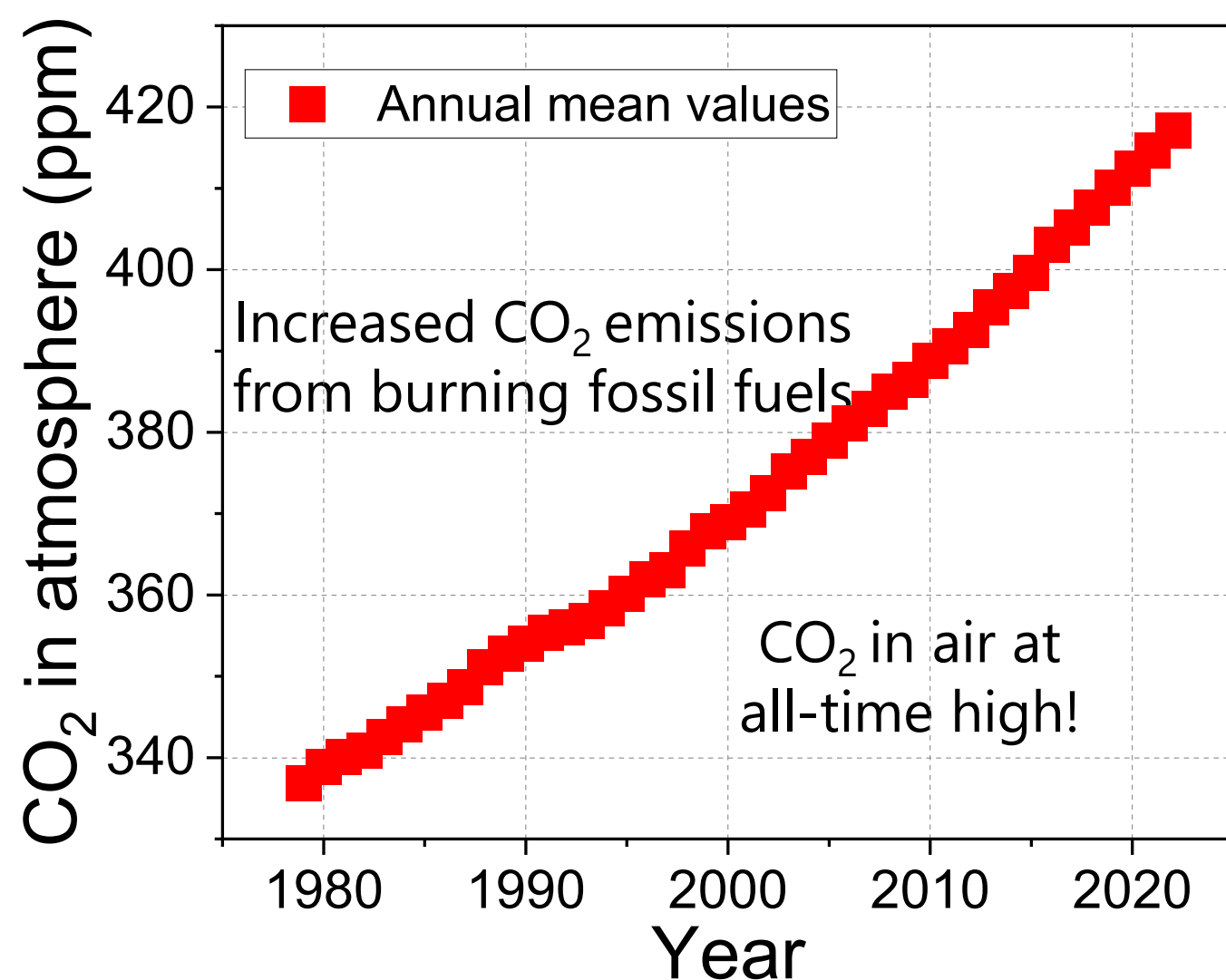
Materials for Flexible Chemical Synthesis in Integrated Carbon Dioxide (CO₂) Capture and Conversion

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THE PROBLEM

36.8 Gt (gigatonnes) of CO₂ in 2022



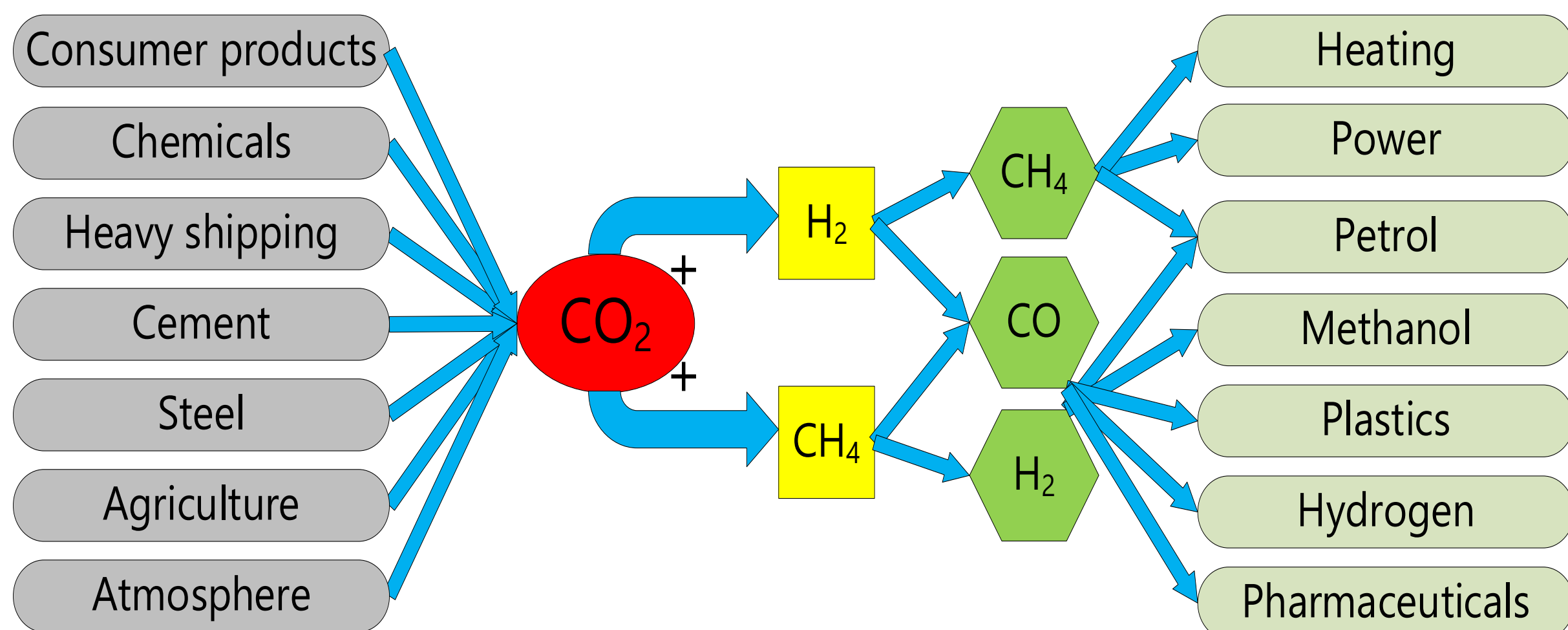
Captured CO₂ costs £40-140 per tonne - very expensive

We need to find a way to produce chemicals **independent of fossil fuels** that can work **anywhere in the world!**

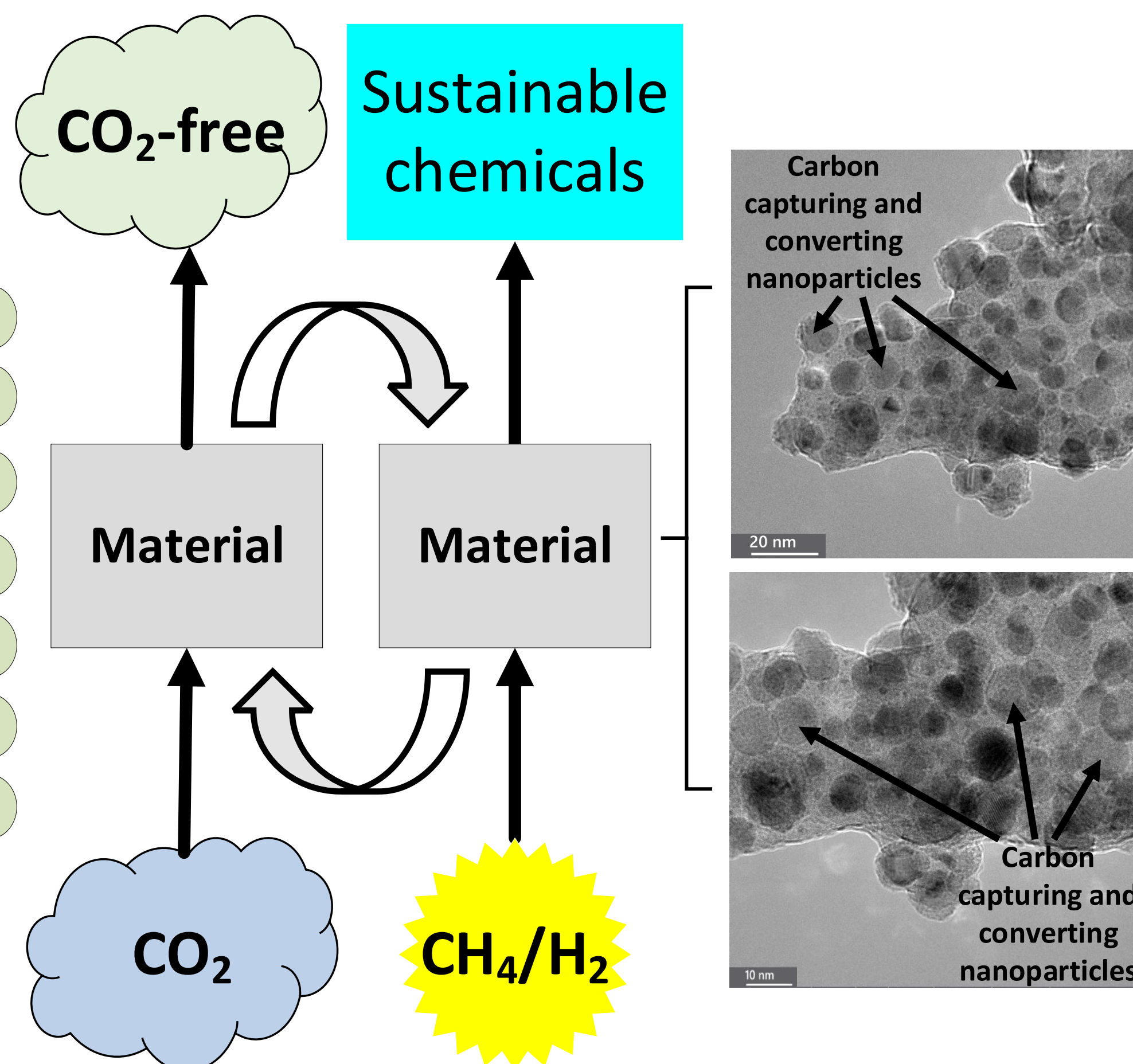


THE SOLUTION

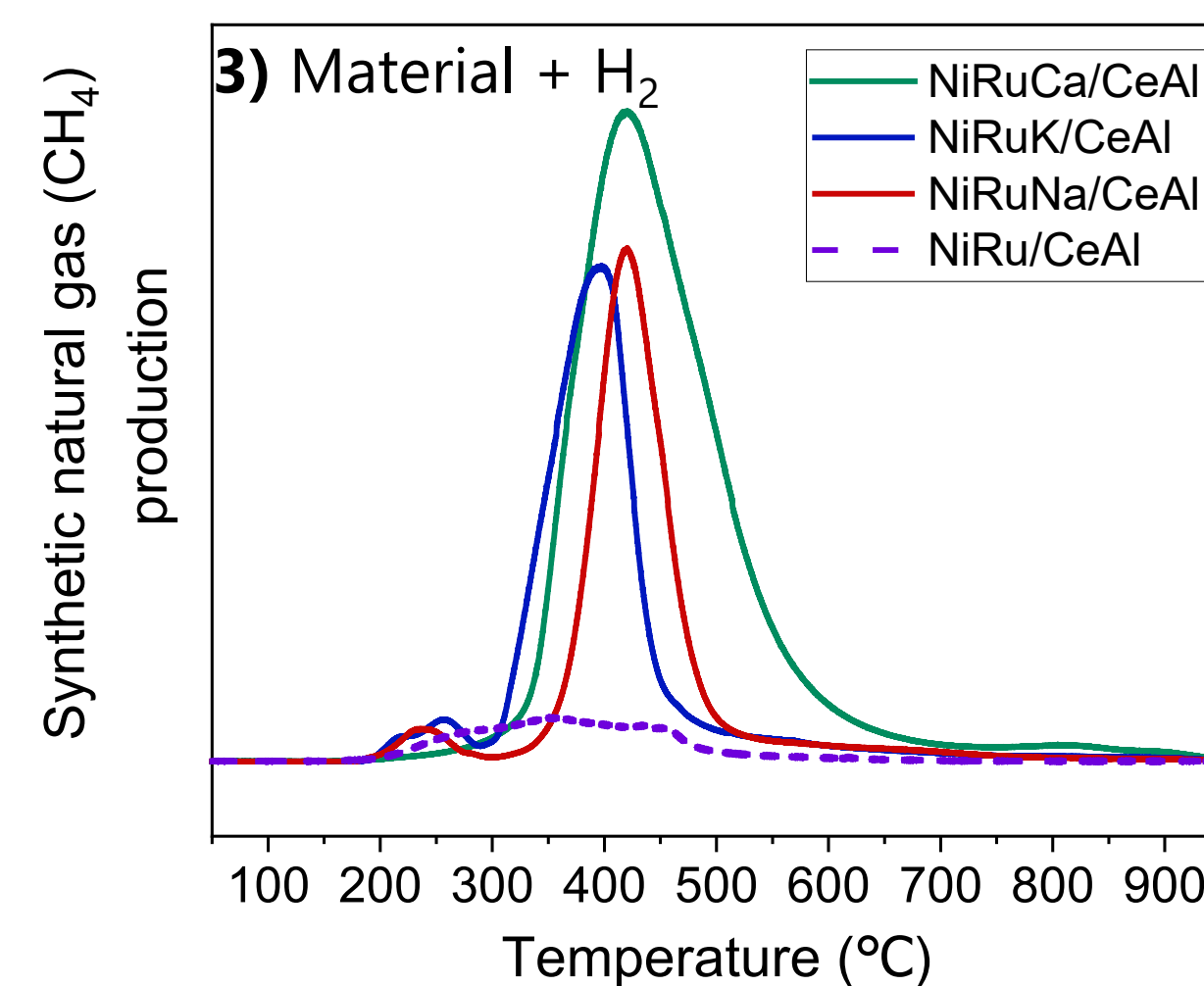
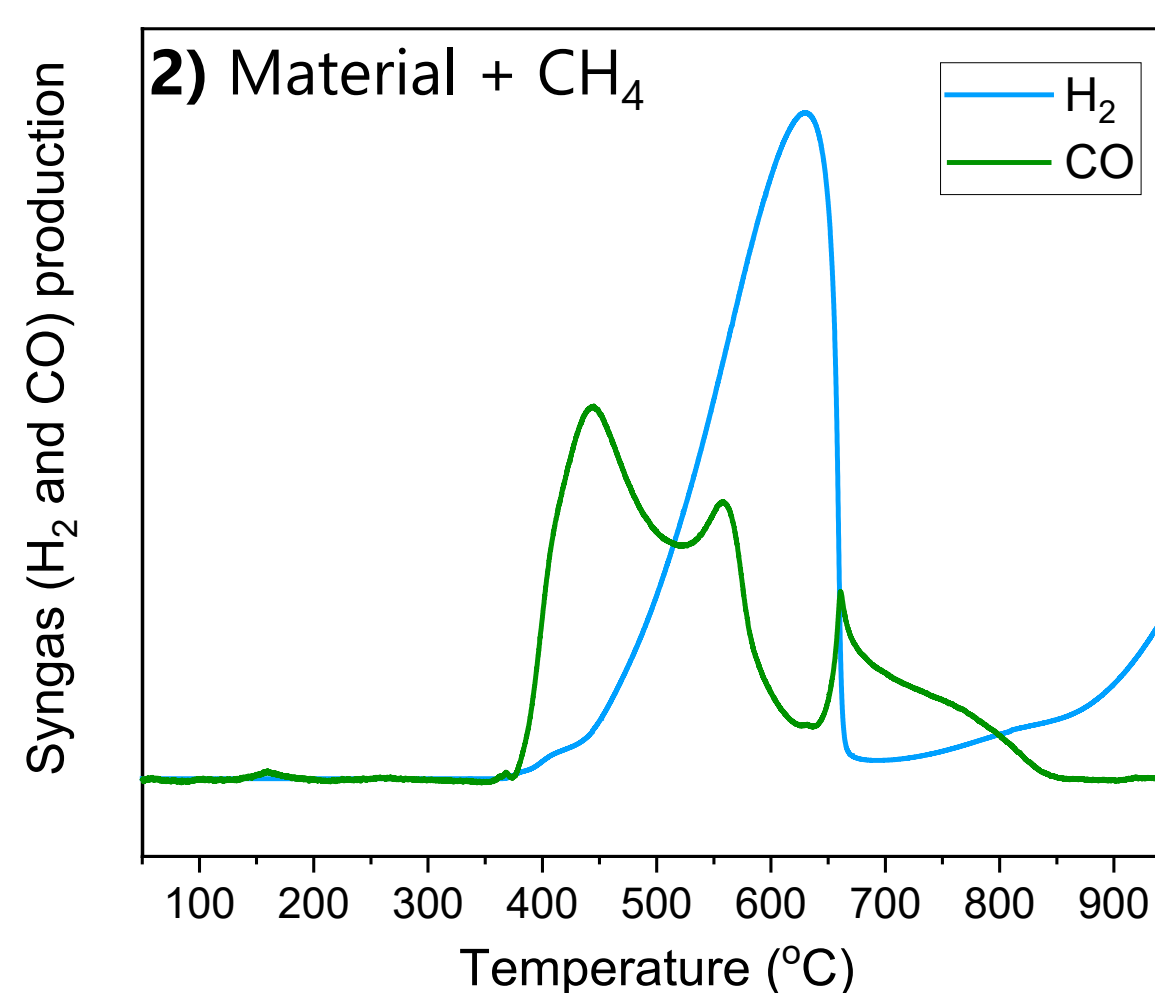
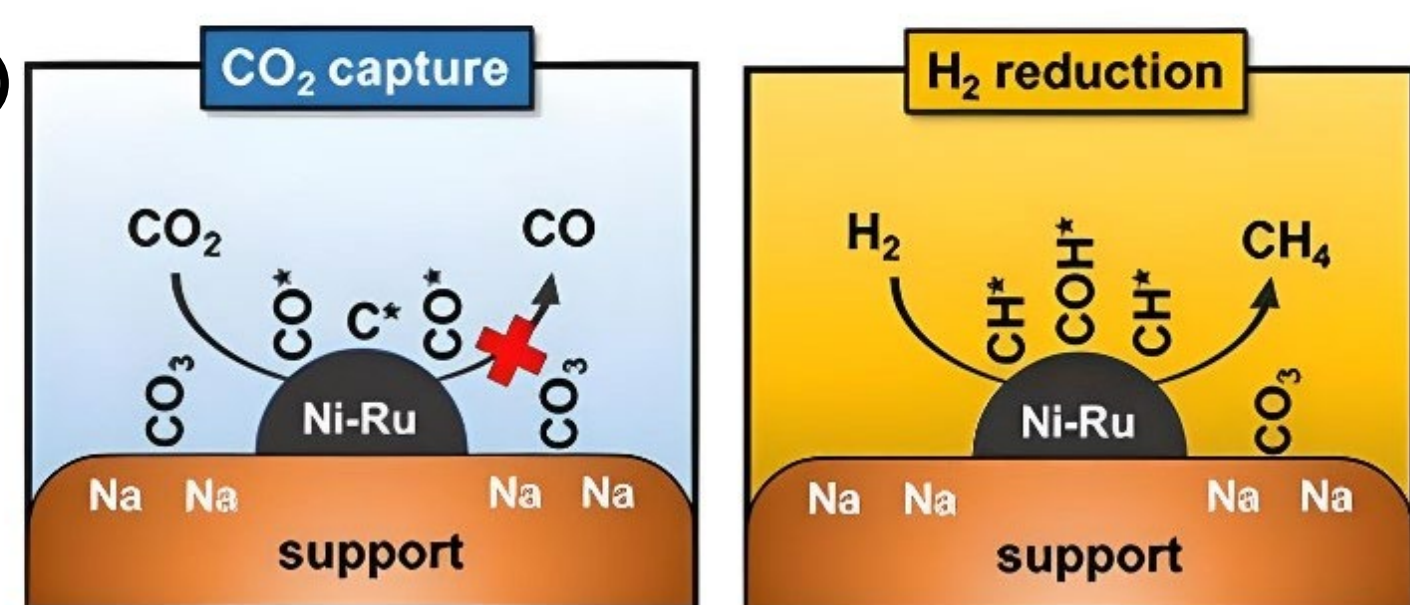
- ✓ **Integrating** the CO₂ capture with the utilisation step, and designing the process to be **flexible**
- ✓ Producing **on-demand chemicals**, using CO₂ as a **carbon source** to turn vice into **virtue**



Our **switchable dual-function materials** capture CO₂ and convert it directly into a product, using hydrogen or hydrocarbons in a cyclic operation.



THE RESULTS



1) By using spectroscopic techniques, we discovered the reaction mechanism of the materials studied in this work.

2) Our materials captured CO₂ directly from the air and converted it into a mixture of carbon monoxide (CO) and hydrogen (H₂), called syngas.

3) Our materials captured CO₂ directly from the air and converted it into synthetic natural gas (CH₄).

TAKE-HOME MESSAGE

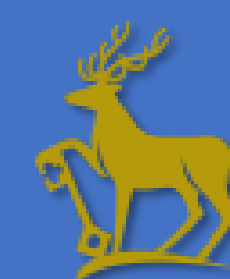
We have designed and patented a material which produces most of the chemicals that our society needs out of every possible source, ranging from thin air to the majority of the heavy emitters, like cement and steel.



References

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- [3] Merkouri, L.P.; Reina, T.R.; Duyar, M.S. *Energy and Fuels* 2021, 35, 19859-19880.
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- [5] Merkouri, L.P.; et. al. *Journal of Materials Chemistry A* 2023, 11, 13209-13216.

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